Apatite crystallites in bone

The characteristics of the wide-angle X-ray reflections from the inorganic component of compact adult bone suggest that this component is present in the form of very small crystallites which are markedly elongated in the direction of the crystallographic c-axis. Calculations from linewidth measurements give the order of size of the crystallites as up to a few hundred Angstroms and a more specific estimate of the length of the long axis from considerations of c-axis reflections alone gives a value of about 200 A¹. Experiments with synthetic apatites show that crystallites of these dimensions produce an intense low-angle X-ray scatter.

Studies of the continuous low-angle scatter from intact bone have indicated the presence in the system of strongly scattering units measuring approximately $200 \text{ A} \times 40$ to $70 \text{ A}^{2,3}$. Such units persist, and scatter even more strongly, after the removal of the organic components from the bone. When the specimen is heated to high temperatures, the size of these scattering units appears to increase, and it is also observed that the definition of the wide-angle reflections is markedly improved, thus indicating a probable increase in the size of the apatite crystallites. In embryonic bone⁴ it has been observed that the initial apatite wide-angle diffraction pattern does not show any preferred orientation (*i.e.* crystallites not well-oriented), but the collagen diffraction pattern is highly oriented. The diffuse low-angle scatter from these specimens shows no orientation.

Thus on the grounds of X-ray diffraction data alone one can conclude that the wide-angle reflections and at least a part of the low-angle scatter probably arise from apatite crystallites. As the low-angle scatter can be interpreted in terms of a single set of particles of uniform shape and size and similar orientation, we suggest that the scatter arises mainly from the apatite crystallites, and that scatter from other particles must either be relatively weak or be of a form similar to that from the apatite. The study of the low-angle scatter from collagen⁵ suggests that this too contains scattering units of a similar form and probably relatively weak. We consider this work to support our earlier contention that there is some close relationship between the organisation of the apatite crystallites and the structure of collagen⁶.

It is to be hoped that the complete picture will eventually be revealed by the electron microscope, but the results so far obtained have been inconclusive.

The recent comments by Caglioti et al. on our conclusions regarding the low-angle scatter from bone emphasise the difficulties encountered in correlating the information obtained from various types of study but contribute little towards resolving these difficulties. The idea that the low-angle scatter comes mainly from holes filled with ossein ignores the independent evidence for the presence of strongly scattering apatite crystallites.

Institutionen for Medecinsk Fysik, Karolinska Institutet, Stockholm (Sweden)

J. B. FINEAN* A. ENGSTRÖM

- ¹ A. TORBORG JENSEN AND E. MÖLLER, J. Dental Research, 27 (1948) 524.
- ² J. B. Finean and A. Engström, Biochim. Biophys. Acta, 11 (1953) 178.
- ³ D. CARLSTRÖM AND J. B. FINEAN, Biochim. Biophys. Acta, 13 (1954) 183.
- ⁴ G. Wallgren, unpublished results.
- ⁵ P. M. COWAN, A. C. T. NORTH AND J. T. RANDALL, Symposia Soc. Exptl. Biol., 9 (1955) 115.
- ⁶ D. Carlström, A. Engström and J. B. Finean, Symposia Soc. Exptl. Biol., 9 (1955) 85.
- ⁷ V. CAGLIOTI, A. ASCENZI AND A. SANTERO, Biochim. Biophys. Acta, 21 (1956) 425.

Received October 29th, 1956

Pyridine nucleotide-dependent reduction of the α -keto acid analogue of lysine to L-pipecolic acid*

Conversion of lysine to pipecolic acid has been demonstrated by tracer studies in plants^{1,2,3} and in the intact rat⁴. That the a-keto acid analogue of lysine may be an intermediate in this conversion is consistent with (a) the observation that the a-amino group rather than the ε -amino group of

 $^{^{\}star}$ Present address: Dept. of Medical Biochemistry and Pharmacology, University of Birmingham.

^{*}This work was supported in part by a research grant from the National Heart Institute, National Institutes of Health, Public Health Service, and by a grant from E. I. du Pont de Nemours and Company.

Abbreviations: DPN, diphosphopyridine nucleotide; TPN, triphosphopyridine nucleotide.